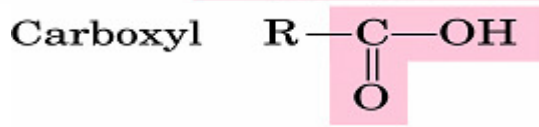
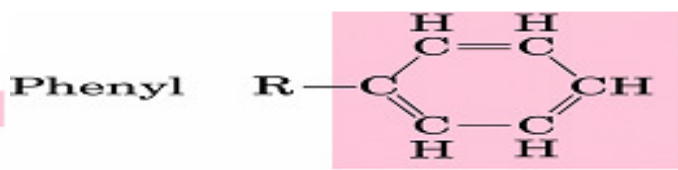
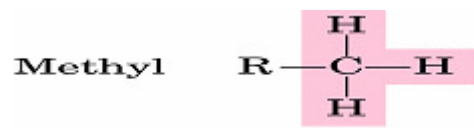
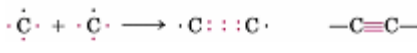
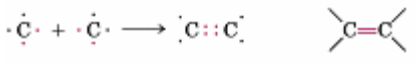
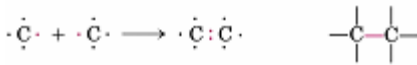
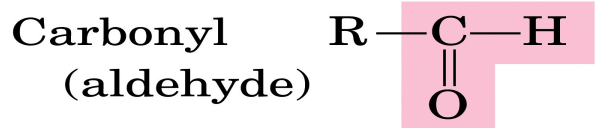
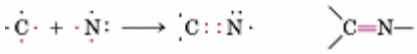
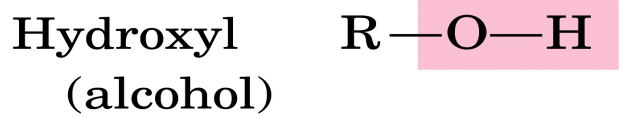
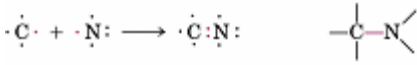
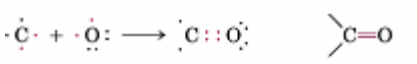
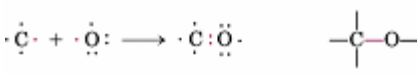
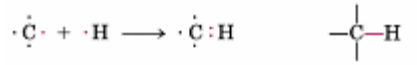
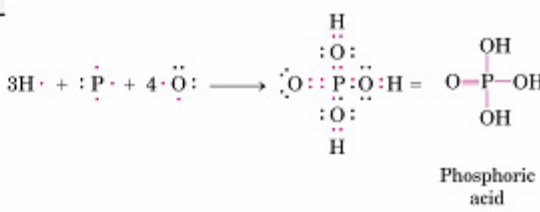
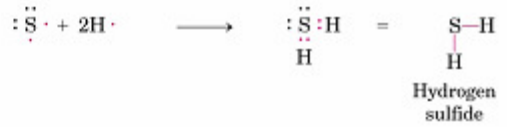
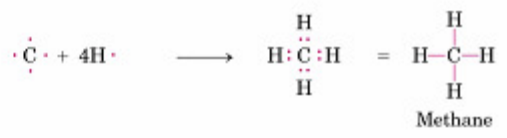
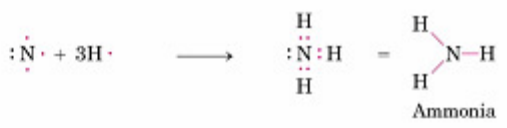
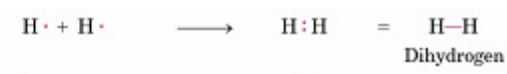
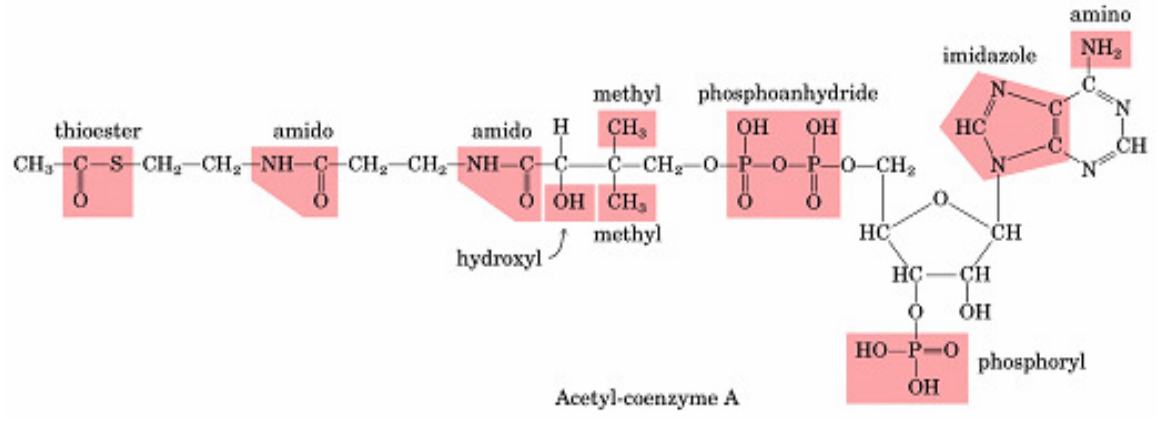
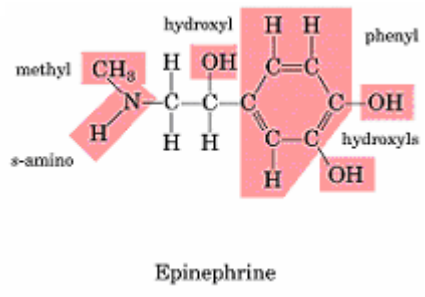
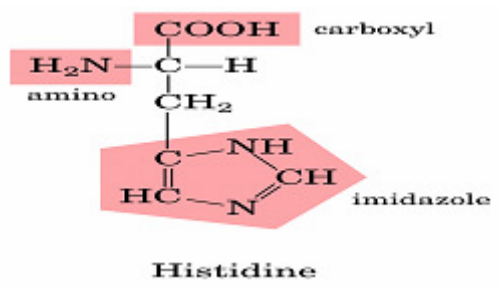
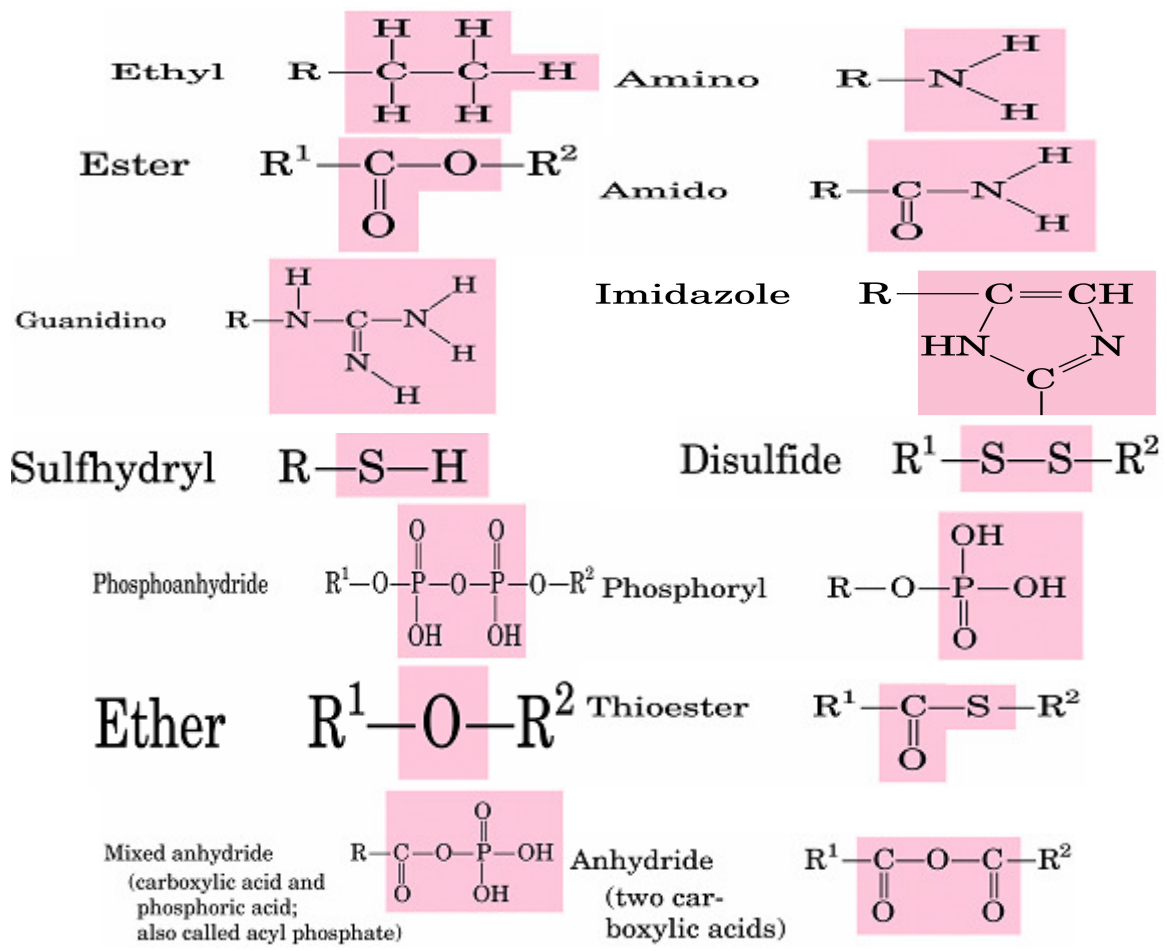
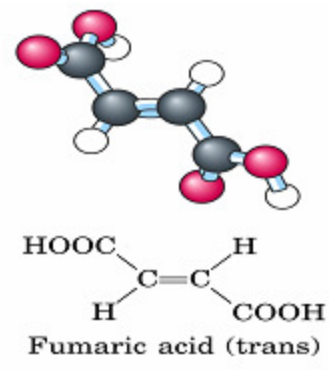
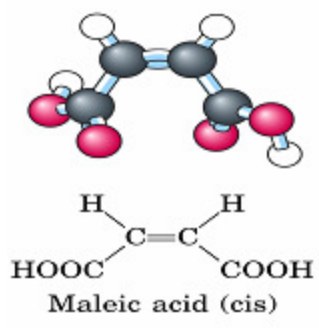


Biomolecules

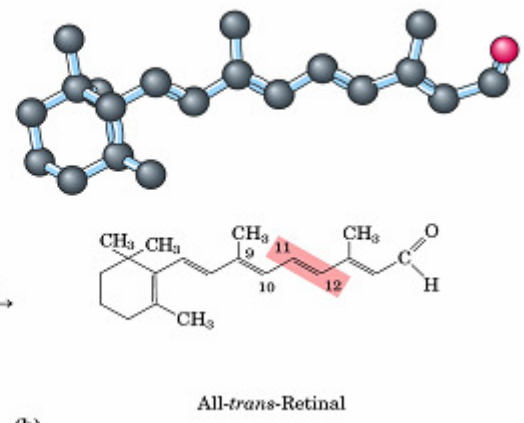
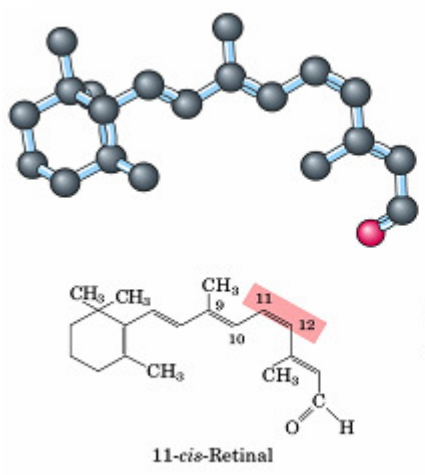
Atom	Number of unpaired electrons (in red)	Number of electrons in complete outer shell
$\cdot\text{H}\cdot$	1	2
$\cdot\ddot{\text{O}}\cdot$	2	8
$\cdot\ddot{\text{N}}\cdot$	3	8
$\cdot\ddot{\text{C}}\cdot$	4	8
$\cdot\ddot{\text{S}}\cdot$	2	8
$\cdot\ddot{\text{P}}\cdot$	3	8



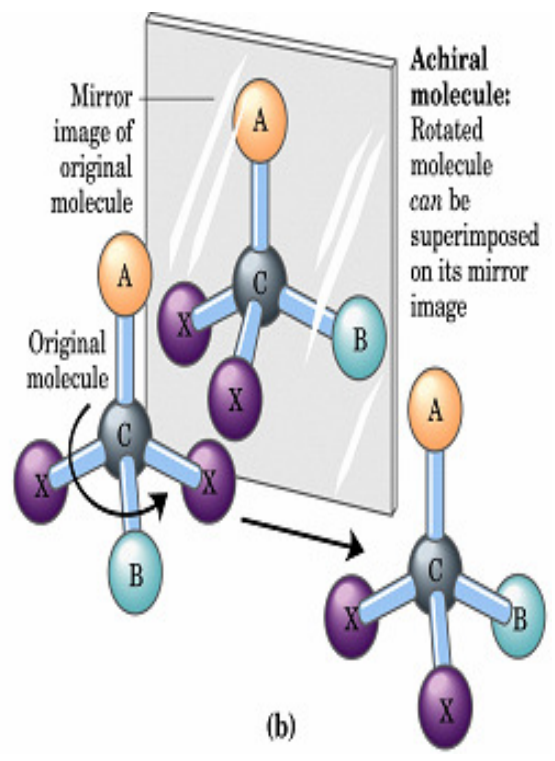
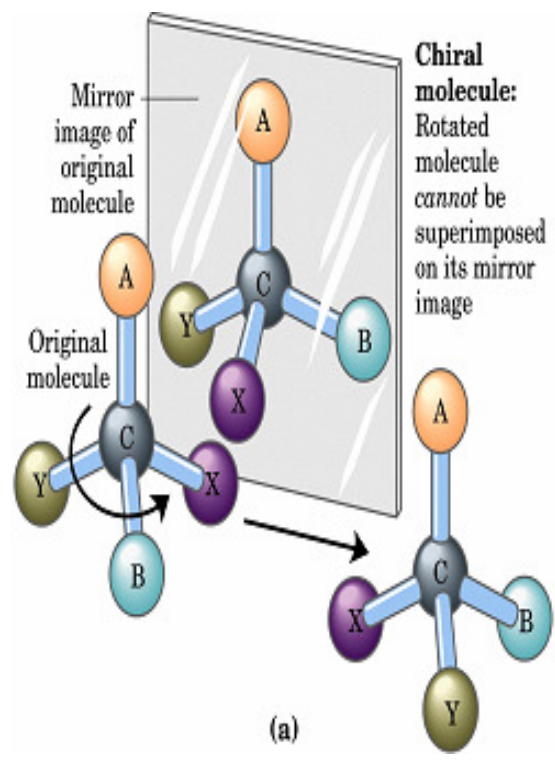


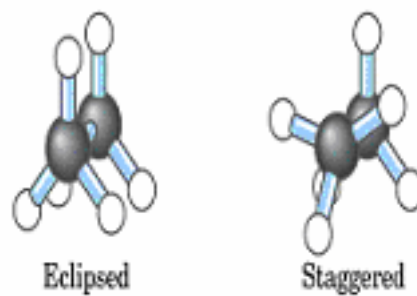
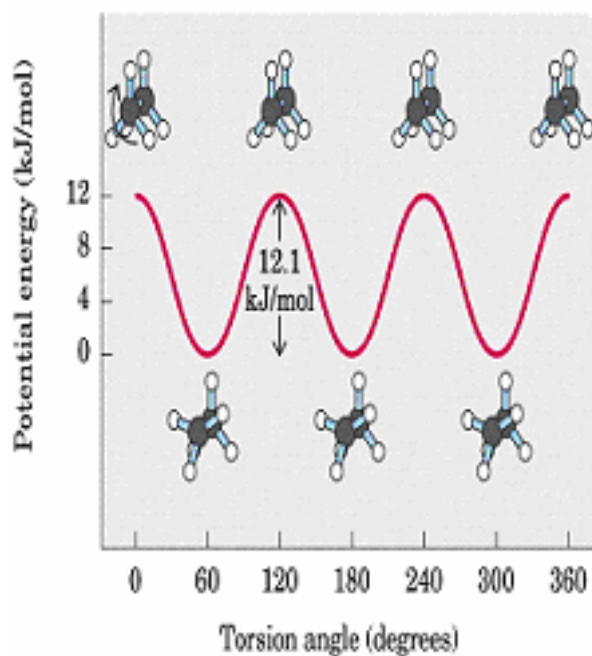
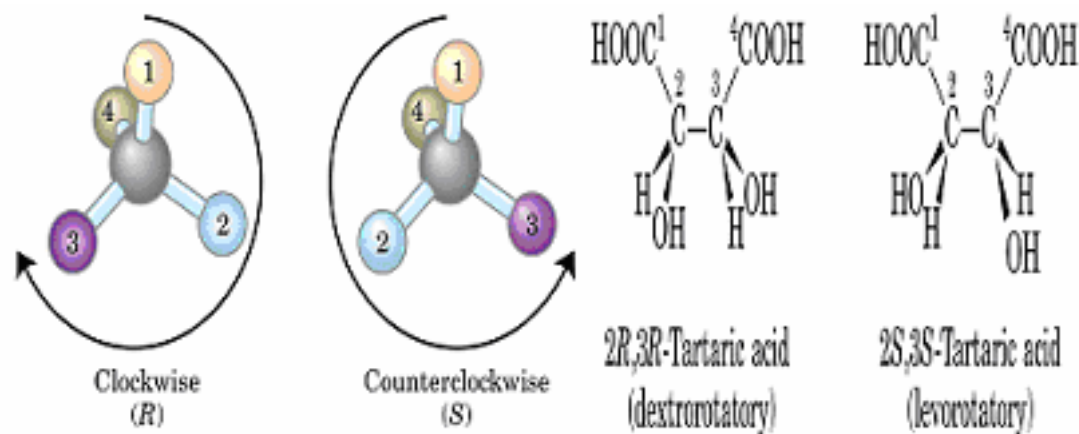
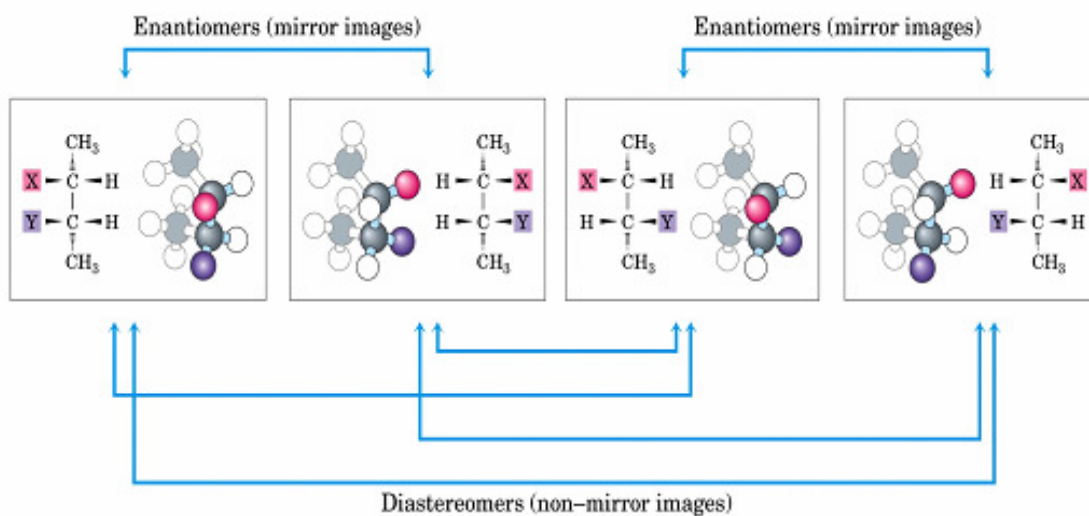


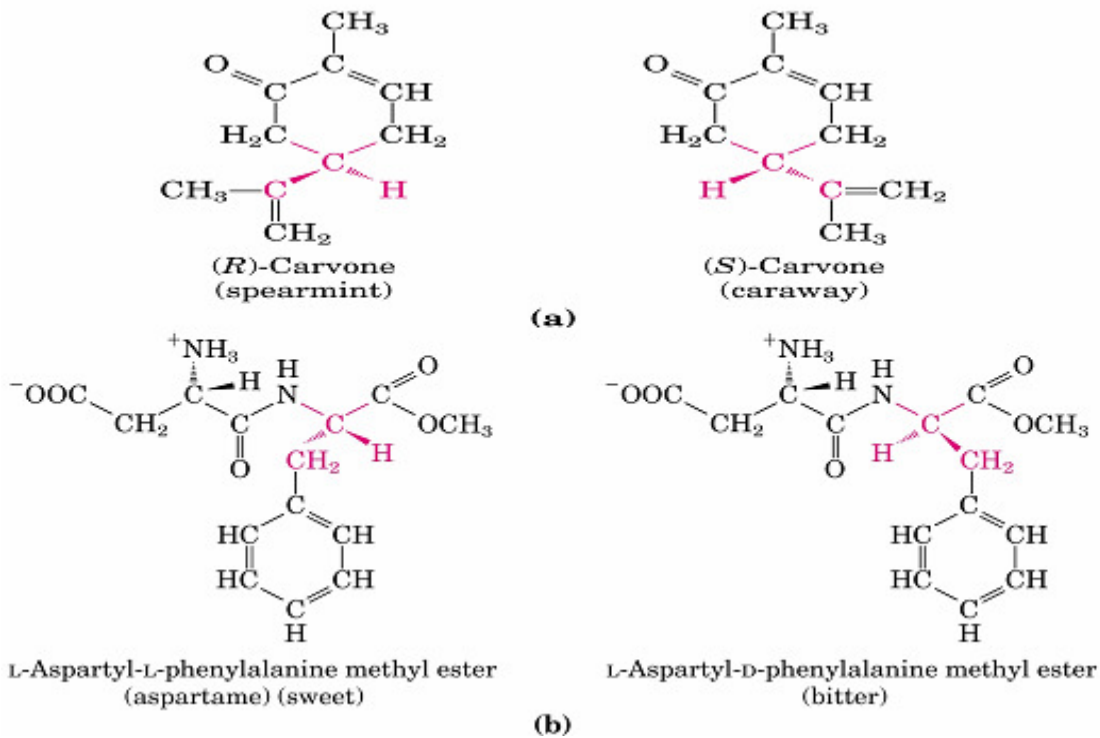
(a)



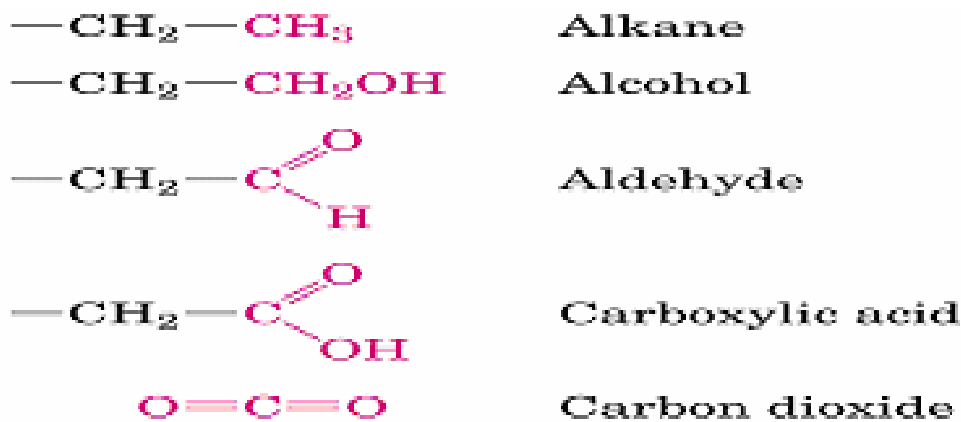
(b)



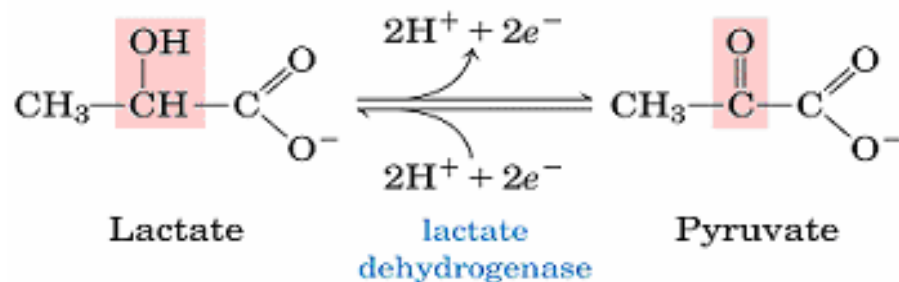




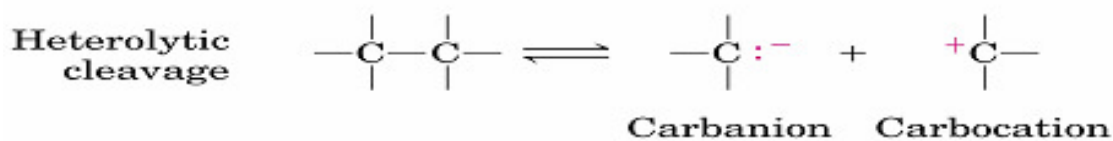
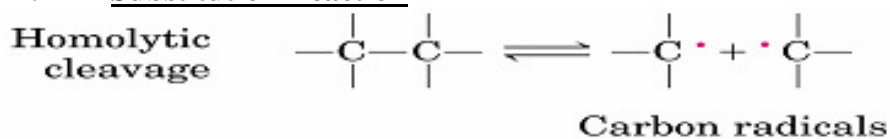
Oxidation States of carbon




1. Redox Reactions



2. Substitution Reaction



Some Functional Groups Active as Nucleophiles within Cells*

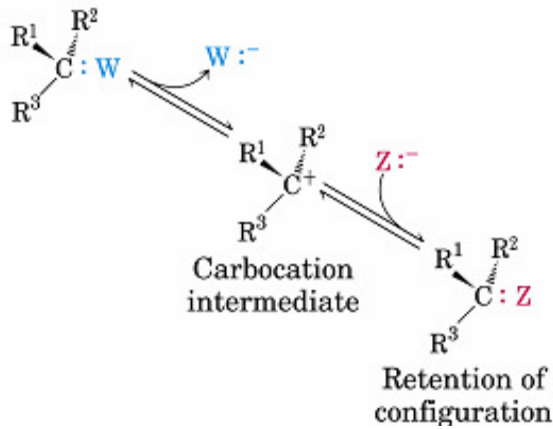
Water	$\text{H}\ddot{\text{O}}\text{H}$
Hydroxide ion	$\text{H}\ddot{\text{O}}\text{:}^-$
Hydroxyl (alcohol)	$\text{R}\ddot{\text{O}}\text{H}$
Alkoxy	$\text{R}\ddot{\text{O}}\text{:}^-$
Sulfhydryl	RSH
Sulfide	$\text{R}\ddot{\text{S}}^-$
Amino	$\text{R}\ddot{\text{N}}\text{H}_2$
Carboxylate	$\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}^-$
Imidazole	
Inorganic orthophosphate	$\text{—O—}\overset{\text{O}}{\parallel}{\text{P}}\text{—OH}$ $\quad \quad \quad \text{O}^-$

*Listed in order of decreasing strength. Weaker nucleophiles make better leaving groups.

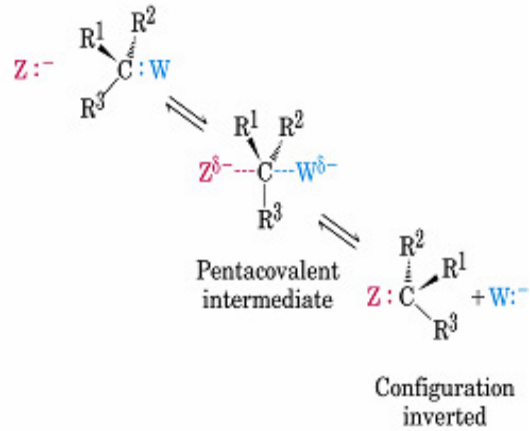
3. Nucleophilic Substitution Reaction



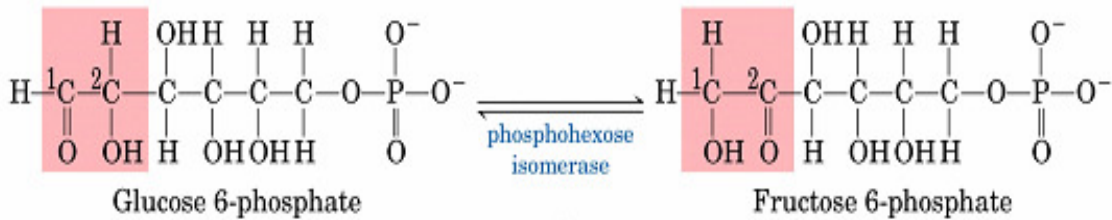
(a) **SN1 reaction**



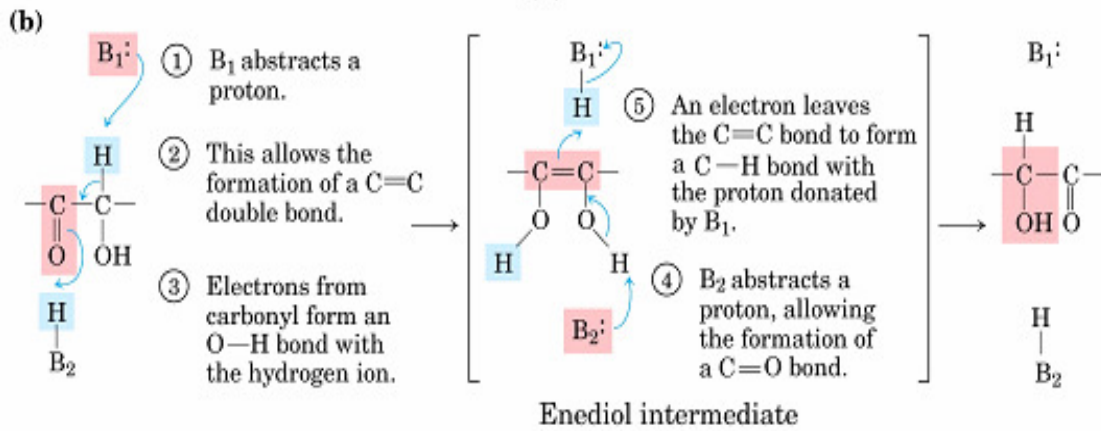
(b) **SN2 reaction**



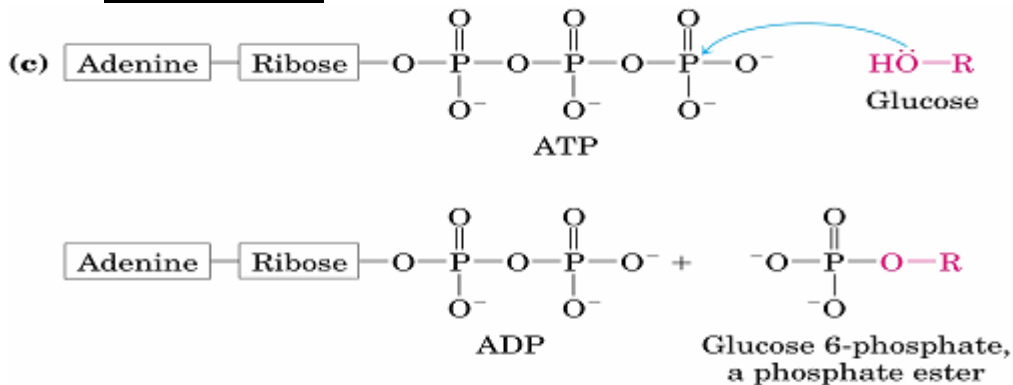
4. **Internal Rearrangement**



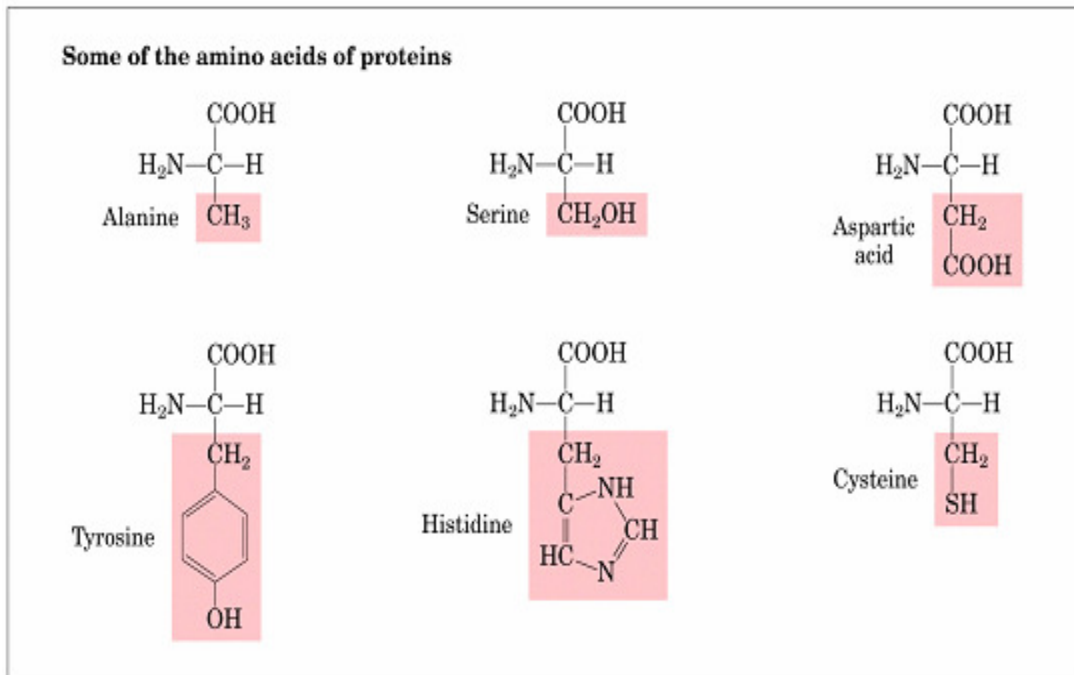
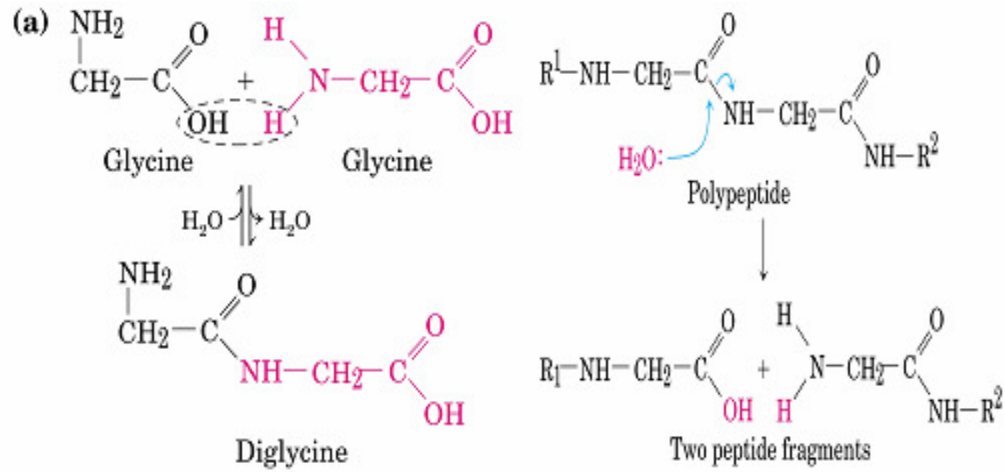
(a)



5. **Group Transfer**

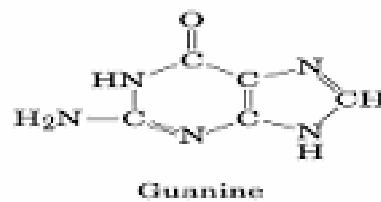
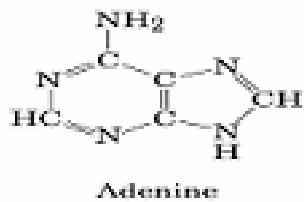
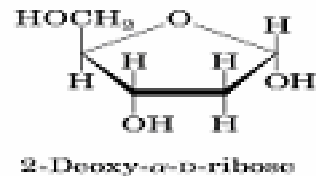
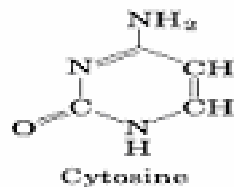
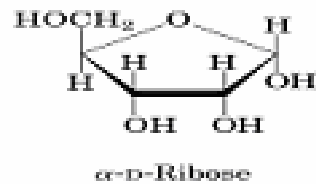
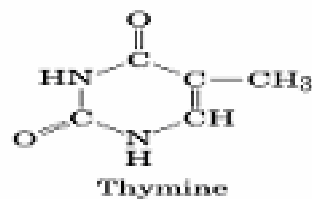
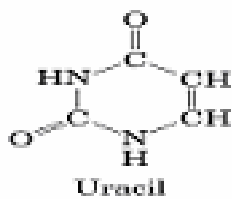


6. Condensation/Hydrolysis reactions.

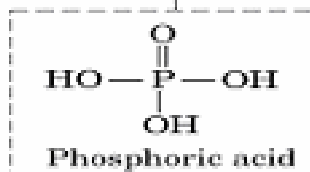
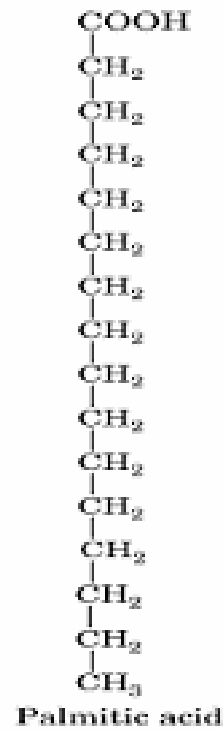
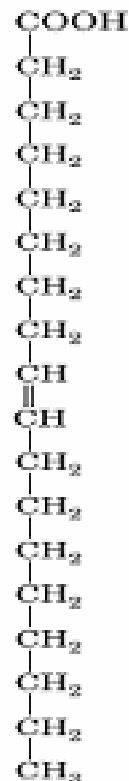
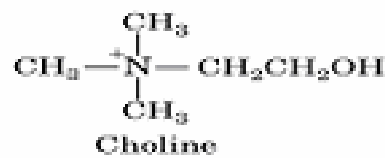


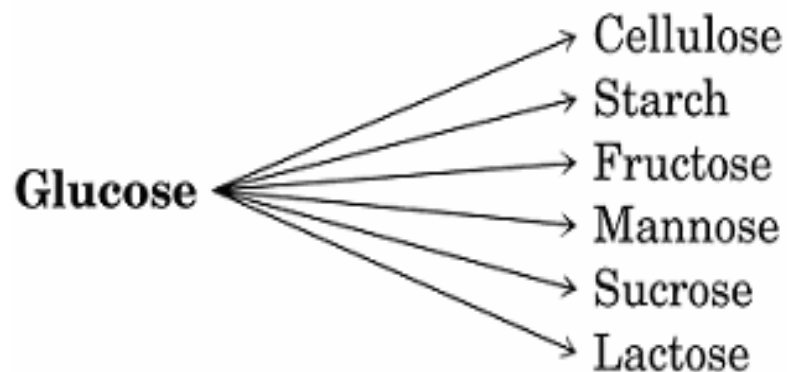
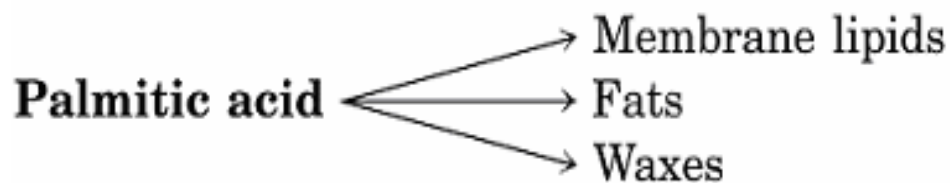
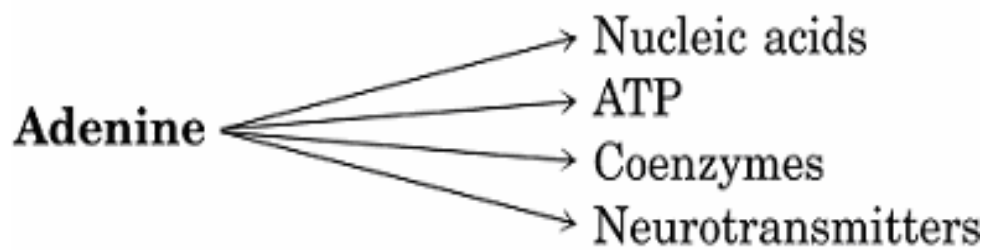
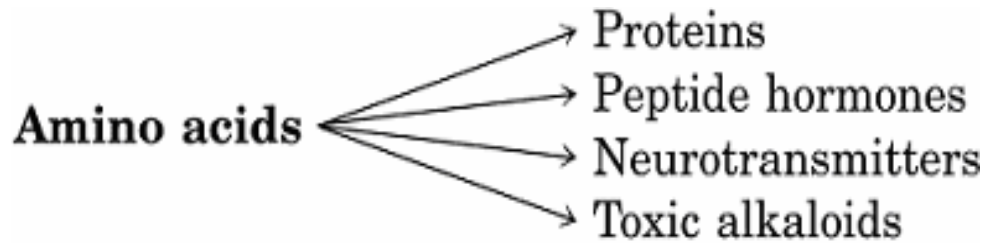
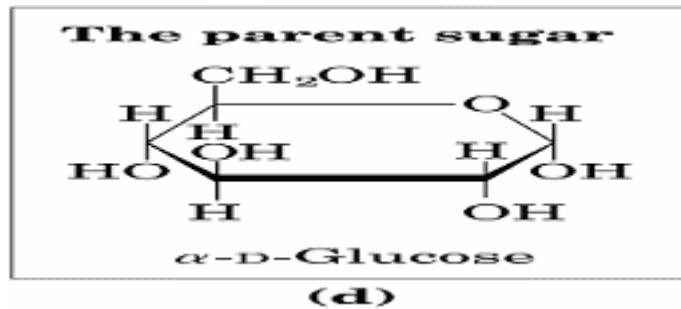
(a)

The components of nucleic acids

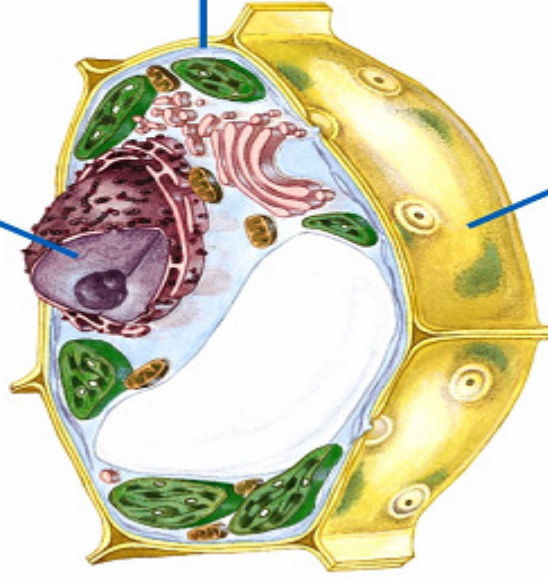


Some components of lipids

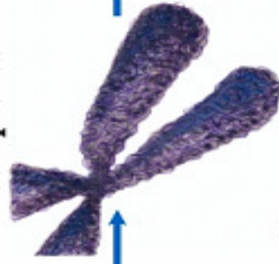




Level 4:
The cell
and its organelles



Level 3:
Supramolecular
complexes



Chromosome

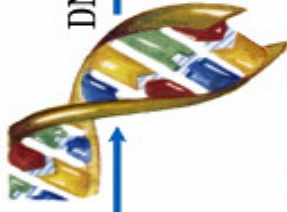


Plasma membrane



Cell wall

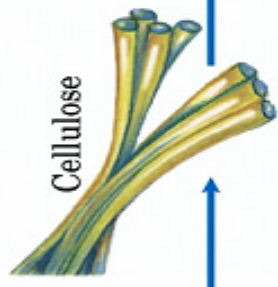
Level 2:
Macromolecules



DNA

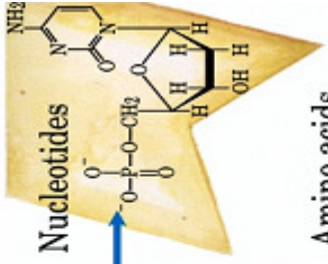


Protein

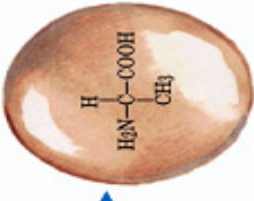


Cellulose

Level 1:
Monomeric units



Nucleotides



Amino acids



Sugars



How did life begin? = PREBIOTIC EVOLUTION

Creation of prebiotic soup, including nucleotides,
from components of Earth's primitive atmosphere



Production of short RNA molecules
with random sequences



Selective replication of self-duplicating
catalytic RNA segments



Synthesis of specific peptides,
catalyzed by RNA



Increasing role of peptides in RNA replication;
coevolution of RNA and protein



Primitive translation system develops,
with RNA genome and RNA-protein catalysts



Genomic RNA begins to be copied into DNA



DNA genome, translated on RNA-protein complex
(ribosome) with protein catalysts